

1. Product and Company Identification

Product Name: Natural Gas Liquids Synonyms: NGLs, Y-grade, C2+

UN Number: 1965

Recommended Use: Feedstock, Fuel

Dominion Energy Questar Pipeline Supplier Address:

333 South State Street

P.O. Box 45433

Salt Lake City, UT 84145-0433

801-324-5111

Chemical Emergency

Phone No.:

801-324-5111

2. Hazards Identification

EMERGENCY OVERVIEW

DANGER!

EXTREMELY FLAMMABLE LIQUID AND VAPOR - MAY CAUSE FLASH FIRE OR EXPLOSION

CONTAINS BENZENE - A KNOWN HUMAN CARCINOGEN MAY BE HARMFULL OR FATAL IF INHALED, SWALLOWED OR ABSORBED THROUGH THE SKIN - MAY BE ASPIRATED INTO THE **LUNGS IF SWALLOWED** MAY CAUSE EYE AND SKIN IRRITATION



Health Hazard Classification:

Carcinogenicity - Category 1A Reproductive Toxicity - Category 2 Acute Toxicity (inhalation) - Category 1 Acute Toxicity (oral) - Category 2 Skin Irritation - Category 2 Aspiration - Category 1

Physical Hazard Classification:

Flammable Liquids - Category 1

Potential Health Effects:

Primary Route of

Inhalation, skin contact

Exposure:

Inhalation: Irritating to the eyes, nose and throat and narcotic at high concentrations.

> May act as a simple asphyxiant. Inhalation of high concentration can cause dizziness, headache, difficulty breathing and loss of consciousness.

Inhalation of the concentrated vapor may cause pulmonary edema.

Eye Contact: Contact with the eyes may cause irritation. Contact with cold gas may

cause frostbite.

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Skin Contact: May cause irritation, reddening and blistering with prolonged exposure.

May be absorbed through the skin. Contract with cold gas may cause

rostbite.

Ingestion: Moderately toxic by ingestion, may cause depression, headache, nausea,

vomiting and swelling of the abdomen. May cause pulmonary edema or

pneumonitis upon aspiration.

Chronic Effects: Long term misuse may cause motor neuropathy and cardiac arrhythmia.

Contains benzene, a human carcinogen (IARC Group 1). Prolonged exposure to benzene is associated with aplastic anemia and myeloid

leukemia.

Aggravated Medical

Conditions:

Respiratory disorders; blood chemistry disorders; skin disorders.

3. Composition/Information on Ingredients

Natural gas liquid is a complex mixture of volatile hydrocarbons, the composition of which varies depending upon the source and processing method of natural gas. Typically, the hydrocarbons range from C3 to C10, with aromatic compounds including benzene, toluene and xylenes.

CHEMICAL	CAS NO.	VOLUME %	
Natural Gas Condensate	68919-39-1	100	
Methane	74-82-8	0 – 2.5	
Ethane	74-84-0	4.5 - 10	
Propane	74-98-6	16 - 25	
Butanes (n-, & iso-)	106-97-8; 78-28-5	22 - 38	
Pentanes (n-, & iso-)	109-66-0; 78-78-4	17 - 28	
<i>n</i> -Hexane	110-54-3	2.5 - 4	
Hexane isomers	Mixture	4.5 - 8	
Heptane isomers	Mixture	5.5 - 10	
Octane isomers	Mixture	1 – 3.5	
C ₉ to C ₁₀ Hydrocarbons	Mixture	0. 5 – 2.5	
Benzene	71-43-2	0.05 - 2	
Toluene	108-88-3	0.05 - 2	
Xylenes	1330-20-7	0.05 – 1.5	

4. First-aid Measures

Inhalation:

If inhaled remove affected person to fresh air, first ensuring your own safety. If experiencing breathing difficulty administer oxygen. If not breathing, administer CPR or artificial respiration. Seek medical attention immediately.

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Ingestion: If swallowed, DO NOT INDUCE VOMITING. If the victim is conscious,

administer water by mouth. Call a physician or poison control center

immediately.

Eye Contact: In the case of eye contact, rinse the eye with plenty of running water for at

least 15 minutes. If frostbite conditions, rinse the eye with tepid water.

Obtain medical assistance.

Skin Contact: Remove contaminated clothing, wash skin with plenty of running water. If

frostbite conditions, wash skin with tepid water. Obtain medical

assistance if irritation persists.

5. Fire-fighting Measurements

Flammable Properties: Flammable liquid

Suitable Extinguishing

Media:

Flame can be extinguished with dry chemical, CO₂ or foam.

Explosion Hazards: Vapors are heavier than air and may travel to a source of ignition and

flash back. Vapors may ignite explosively. Liquid may float and ignite on

the surface of water.

Hazardous Combustion

Products:

Irritating gases of oxides of carbon including carbon dioxide and carbon

monoxide.

Special PPE & Precautions

for Fire-fighters:

Evacuate the area and fight the fire from a safe distance. Water may be ineffective to extinguish the fire, but should be used to keep surroundings and containers cool. If a leak or spill has not ignited, use water spray to disperse the vapors and to protect personnel attempting to stop a leak. Water spray may be used to flush spills away from areas of potential ignition. Divert run-off water away from sewers and waterways. Wear approved respiratory equipment and full protective equipment as indicated

for fighting fire.

6. Accidental Release Measures

Personal Precautions: All sparks, flames and sources of ignition must be restricted from the

area, ventilate if indoors by opening doors and windows. Evacuate and

clear a safe area.

Wear self-contained breathing apparatus and protective clothing where

warranted.

Review Section 5, Fire-fighting Measures, before proceeding with

containment and cleanup procedures.

Environmental Precautions: Avoid washing, draining, or directing material into sewers and drainage.

Containment and Cleanup: Contain spill immediately in smallest possible area. Recover as much of

the product as possible by such methods as vacuuming and return liquid to an appropriate container. Soak up residual liquid by using absorbent materials such as vermiculite, sand or clay and placing in a container along with other contaminated material, soils or debris for ultimate

disposal.

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7. Handling and Storage

Safe Handling: Prevent eye and skin contact; do not breathe the mist or vapors. Wash

hands after handling. Use in a well-ventilated area. Ground and bond all lines, containers and equipment used with natural gas condensate to prevent static sparks. Do not weld, cut, or grind on containers which have

been emptied; flammable vapors and residues may remain.

Safe Storage: Store in approved, labeled containers in a ventilated area; ensure

containers are tightly closed. Keep containers away from flame, sparks and excessive temperatures. Keep away from incompatible chemicals including

strong oxidizers. Comply with all federal, state and local regulatory requirements for handling, storage, transfer and disposal of flammable

liquids.

8. Exposure Controls/Personal Protection

OUEMOAL	OCCUPATIONAL EXPOSURE LIMIT (OEL) ¹			
CHEMICAL	OSHA	ACGIH	NIOSH	
Methane	Not applicable	1000 ppm ²	Not applicable	
Ethane	Not applicable	1000 ppm ²	Not applicable	
Propane	1000 ppm	1000 ppm ²	1000 ppm TWA 2100 ppm IDLH	
Butanes (n-, & iso-)	Not applicable	1000 ppm ³	800 ppm ⁴ TWA	
Pentanes (n-, & iso-)	1000 ppm⁵	600 ppm	120 ppm ⁵ 610 ppm ⁵ STEL 1500 ppm ⁵ IDLH	
<i>n</i> -Hexane	500 ppm	50 ppm (skin)	50 ppm 1100 ppm IDLH	
Hexane isomers		500 ppm 1000 ppm STEL	100 ppm 510 ppm STEL	
Octane isomers	500 ppm ⁶	400 ppm 500 ppm STEL	85 ppm ⁶ 440 ppm ⁶ STEL 750 ppm ⁶ IDLH	
Octane isomers	500 ppm ⁷	300 ppm	75 ppm ⁷ 385 ppm ⁷ STEL 1000 ppm IDLH	
C ₉ – C ₁₀ Hydrocarbons		200 ppm ⁸	200 ppm ⁹	
Benzene	1 ppm 5 ppm STEL	0.5 ppm (skin) 2.5 ppm STEL	0.1 ppm 1 ppm STEL 500 ppm IDLH	
Toluene	200 ppm 300 ppm C 500 ppm Peak	20 ppm	100 ppm 150 STEL 500 ppm IDLH	
Xylenes	100 ppm	100 ppm 150 ppm STEL	100 ppm 150 STEL 900 ppm IDLH	

¹OELs are 8-hour time weighted average exposure levels unless otherwise indicated as IDLH (Immediately Dangerous to Life and Health), STEL (15 minute Short Term Exposure Limit or Ceiling Limit), C (Ceiling Limit) or Peak (maximum ten minute concentration above the Ceiling Limit). Limits with a "skin" notation indicate that the substance can be

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absorbed through the skin.

Personal Protective Measures and Controls:

Eye Protection: Avoid eye contact. Chemical goggles or face shield if there is a potential for

sprays, mists or splashes.

Skin and Body protection: Avoid skin contact. Wear chemical protective gloves. Consult manufacturer

for glove specifications. Where splashing is possible, wear protective coveralls. Remove immediately if contaminated and launder before re-use.

Respiratory Protection: Where necessary to maintain exposure levels below the Occupational

Exposure Limits in Section 8, select appropriate NIOSH approved respiratory protection. Proper respirator selection should be determined by adequately trained personnel, and based on the contaminant(s), the potential exposure, and the published respirator protection factor.

Hygiene measures: Observe good hygiene practices. Wash after handling the material, and

before eating, drinking or smoking. Wear clean protective clothing; contaminated clothing should be promptly removed for laundering or

replacement.

Engineering and Ventilation

Controls:

Where applicable, adequate general or local exhaust ventilation should be used to maintain airborne concentrations below occupational exposure levels, to prevent the formation of explosive atmospheric concentrations,

and to prevent the displacement of oxygen in confined areas.

9. Physical and Chemical Properties

Appearance & Odor: Liquefied gas with a petroleum odor/Colorless

Physical Form: Liquid

Flash Point: As low as -50° F (-46° C)

Boiling Point: $< -140^{\circ} \text{ F (-95^{\circ} C)}$

Flammability Limits in Air: LEL < 1 %

UEL = 8.0 %

Vapor Density: > 1

Specific Gravity: 0.65 - 0.75

Solubility in Water: Insoluble

Percent Volatile by Up to 100

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²As an aliphatic hydrocarbon gas (C₁ to C₄)

³Applies to n-Butane as an aliphatic hydrocarbon gas

⁴A NIOSH REL of 800 ppm as a time weighted average exposure has been established for both butanes listed.

⁵OEL applies to *n*-Pentane.

⁶OEL applies to *n*-Heptane.

⁷OEL applies to *n*-Octane.

⁸OEL applies to all isomers of Nonane.

⁹OEL applies to *n*-Nonane.





Volume:

Vapor Pressure: 150 – 200 psia (Reid VP) @ 100°F (38° C)

Autoignition Temperature As low as 450° F (232° C)

10. Stability and Reactivity

Stability: Stable under normal storage and handling conditions.

Conditions to Avoid and Incompatible Products:

Strong oxidizers, sources of heat or ignition. Prevent vapor accumulation.

Hazardous Decomposition

Products;

Incomplete combustion may release carbon monoxide, carbon dioxide and

smoke (non-combusted hydrocarbons).

Hazardous Polymerization: None.

11. Toxicological Information

	INHALATION	ORAL/INGESTION	EYE	SKIN	REPRODUCTIVE
<i>n</i> -Butane	Rat LC ₅₀ 658 gm/m ³ /4H				
<i>n</i> -Pentane	Rat LC ₅₀ 364 gm/m ³ /4H	Mouse LD ₅₀ IV 446 mg/kg			
Iso-Pentane	Mouse LC _{Lo} 419 g/m3/2H				
<i>n</i> -Hexane	Human TCLo 190 ppm/8W (PNS)	Rat oral LD ₅₀ 28,710 mg/kg	Rabbit - Irritation		Inh Rat TCLo 10,000 pp,/17 H reproductive Inh Rat TCLo 5000 ppm Teratogenic
<i>n</i> -Heptane	Human TCLo 1000 ppm/6M (CNS)	Mouse iv LD ₅₀ 222 mg/kg			
Octane	Rat LC ₅₀ 118 g/m3/4H	Mouse iv LDLo 428 mg/kg			
Nonane	Rat LC ₅₀ 3200 ppm/4H	Mouse iv LD ₅₀ 218 mg/kg			
Benzene	Human TC 150 ppm/15M/8Y I (carcinogenic and blood effects) Human 10 mg/m3/11Y I	Human oral LDLo 50 mg/kg	Rabbit 2 mg/24 H Severe	Rabbit 15 mg/24H mild	
Toluene	Human TCLo 100 ppm CNS	Human LDLo 50 mg/kg	Human 300 ppm	Rabbit 435 mg Mild	
Xylenes	TCLo 200 ppm Eye pulmonary LCLo 10,000 ppm/6H	Human LDLo 50 mg/kg	Human 200 ppm	Rabbit 500 mg/24H mod	

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Summary Comments:

Acute: Mildly toxic and narcotic by inhalation or skin absorption, may cause irritation to the respiratory tract and cardiac arrhythmia. Liquid is irritating to the skin and may cause blistering. If ingested, aspiration causes severe lung irritation, coughing and pulmonary edema.

Chronic: Exposure to aliphatic compounds may cause peripheral neuropathy. Chronic exposure to aromatic hydrocarbons may cause leukemia (benzene), aplastic anemia (benzene), and dysfunction of the kidney and liver.

12. Ecological Information

Ecotoxicity: Mixture is harmful to aquatic organisms; the median threshold limit (TL_m) is

less than 100 ppm.

Persistence and degradability: Mixture is expected to degrade naturally.

Bioaccumulation potential: None.

Mobility: The mixture will float on water with loss to air through volatilization, may

foul shorelines. In soil, the mixture is expected to passively volatilize in air.

13. Disposal Considerations

Waste Classification If disposed of in a container, may be defined as a RCRA hazardous waste

by the characteristic, "ignitability" (D001) and benzene (D018). Refer to

Sections 7 and 8 for the safe handling and storage precautions.

All disposal activities must comply with federal, state, and local regulations.

14. Transport Information

When shipping at a vapor pressure > 300 kPa (43.5 psia) at 50° C, the following shipping information applies:

UN Number: UN1965

UN Proper Shipping Name: UN1965, Hydrocarbon gas mixture, liquefied, n.o.s., 2.1

Hazard Class: 2.1

DOT Shipping Label FLAMMABLE GAS

Emergency Response Guide Number: 115

15. Regulatory Information

United States Regulations:

CERCLA Hazardous Substance List

(40 CFR 302.): Benzene (CAS 71-43-2): RQ 10 lbs.

SARA Title III, Section 311: Acute: Yes

Chronic: Yes
Fire: Yes
Pressure: Yes
Reactive: No

CWA, Section 311: Benzene TSCA: None

DOT: 49 CFR Parts 191-192 OSHA 29 CFR 1910.1028, Benzene

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16. Other Information

NFPA: Health Hazard 2 Fire Hazard 4 Instability 0 Special Hazard

HMIS: Health Hazard 2 Flammability 4 Physical Hazard 0 Personal Protection

Date of Issue: May 2013

Abbreviations and Acronyms:

ACGIH American Conference of Governmental Industrial Hygienists

CERCLA Comprehensive Environmental Response, Compensation, & Liability Act

DOT U.S. Department of Transportation

HMIS Hazardous Materials Information System

IARC International Agency for Research on Cancer

IDLH Immediately Dangerous to Life

NIOSH National Institute of Occupational Safety and Health

NFPA National Fire Protection Association

OSHA Occupational Safety and Health Administration

ppm parts per million

SARA Superfund Amendments & Reauthorization Act

STEL Short Term Exposure Limit (typically a 15-minute time weighted average)

TSCA Toxic Substances Control Act

TWA Time Weighted Average (typically 8 hours)

Disclaimer: While proper care has been taken in the preparation of this Safety Data Sheet, this information is provided without warranty. Each individual utilizing this document should make an independent determination of the methods to be used to protect the public, workers and the environment.

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